

# Automated Solid Phase Extraction (SPE) of 10 Perfluorinated Compounds (PFCs) from Tap Water



## **APPLICATION NOTE AN1029**

### **ABSTRACT**

Perfluorinated compounds (PFCs) are a family of molecules consisting of varying lengths of fluorocarbon chains with a functional group such as carboxylic or sulfonic acid attached. Since 2009, PFOs have been classified as POPs (Persistent Organic Pollutants) and the International Agency for Research on Cancer (IARC) classified PFOA as a Group 2B possible carcinogen. The analysis of PFCs can be complex due to their presence in multiple items used in the analytical workflow.

This application note describes the automation of the isolation of 10 PFCs from tap water prior to their analysis by LC/MS using a specific SPE polymer AttractSPE™ WAX-PFC and the Gilson 274 ASPEC\* Large Volume (LV) System. The results show good reproducibility of the method without any contamination from the extraction system.

This collaboration study was performed jointly by Gilson, Inc. and Affinisep.

### **KEYWORDS**

Gilson 274 ASPEC® Large Volume System, TRILUTION® LH Liquid Handling Software, Solid Phase Extraction (SPE), Perfluorinated compounds (PFCs), AttractSPE™ WAX-PFC.

### **INTRODUCTION**

Perfluorinated compounds (PFCs) are a family of molecules consisting of varying lengths of fluorocarbon chains with a functional group such as carboxylic or sulfonic acid attached. To achieve the concentration limit, defined by EU Water Framework Directive and the EPA for perfluorinated compounds in water, it is highly recommended to use solid phase extraction (SPE) to concentrate the sample prior to mass spectrometry analysis.





Figure 1
Gilson 274 ASPEC\* Large Volume System







# Table 1 List of the tested PFCs sorted from the smallest to the largest.

Compound	Chemical Composition	CAS Number
Perfluorobutanoic acid (PFBA)	F F F F	375-22-4
Perfluoropentanoic acid (PFPeA)	F F F OH	2706-90-3
Perfluorohexanoic acid (PFHxA)	F F F F F	307-24-4
Perfluoroheptanoic acid (PFHpA)	F F F F F OH	375-85-9
Perfluorooctanoic acid (PFOA)	F F F F F F F	335-67-1
Perfluorononanoic acid (PFNA)	F F F F F F F F F F F F F F F F F F F	375-95-1
Perfluorodecanoic acid (PFDA)	F F F F F F F F F F F F F F F F F F F	335-76-2
Perfluorobutanesulfonic acid (PFBS)	F F F F Q Q F S OH	375-73-5
Perfluorohexane sulfonic acid (PFHxS)	F F F F OH	355-46-4
Perfluorooctanesulfonic acid (PFOS)	F F F F F F F O C	1763-23-1

This application note describes the analysis of 10 perfluorinated compounds using a specific **AttractSPE™ WAX-PFC**. This cartridge chemistry is dedicated specifically to perfluorinated compounds. The cartridges are used in the Gilson **Gilson 274 ASPEC® Large Volume System** (Figure 1) to purify and concentrate the sample prior to analysis by LC-MS/MS.

The analysis of PFCs can be challenging. They are common material components in analytical systems. Their presence in HPLC systems and solvents, sampling bottles, valves, transfer tubing, etc. can lead to significant chance of sample contamination and high PFCs background, masking trace levels of PFCs from detection and/or leading to false positives.

This application ultimately shows the absence of potential PFCs contamination through the Gilson 274 ASPEC® Large Volume System with high recoveries and reproducibility.

# PRECAUTIONS FOR THE ANALYSIS OF PFCs AND RELIABILITY OF THE METHOD

### PRECAUTIONS FOR SAMPLE HANDLING AND PREPARATION

EPA 537-1<sup>[1]</sup> method describes numerous precautions for sample collection, preservation, storage, analysis, and glassware treatment. It also prohibits the use of materials that can cause accidental contamination, ultimately resulting in cleaner analyses.

For this study, Gilson HDPE sample bottles and Gilson PIPETMAN® DIAMOND Tips were used for sample handling and for standard preparation. The **Gilson 274 ASPEC® Large Volume System** was equipped with PEEK tubing to avoid the use of any fluoropolymer, such as PTFE or FEP, to reduce contamination risk during the sample preparation process.

Cartridges, and the filters and resins contained in them, can also be sources of contamination. For this reason, AFFINISEP has developed a specific **AttractSPE™ WAX-PFC** dedicated to the isolation of perfluorinated compounds. AFFINISEP **AttractSPE™ WAX-PFC** cartridges will not contaminate any sample with additional PFCs.

To reduce contamination risk, best practice indicates that all sampling containers are discarded after use, nitrile gloves are preferred over other glove materials, the nitrile gloves are often replaced, and all glass or fluoropolymer containers are avoided to prevent surface adsorption of PFCs or contamination from container material.

### PRECAUTIONS TO IMPROVE HPLC METHOD RELIABILITY

HPLC devices often contain PTFE parts and tubing. This, coupled with potential traces of PFCs or other fluorine containing material in solvents, can hinder analysis at low concentrations. The fluorine containing materials that are released from the HPLC system tend to build-up at the front of the column creating interference. To minimize this interference, all PTFE or any other fluorinated or perfluorinated polymers are replaced with non-fluorine containing materials. Also, despite the difficulty and added cost, all solvents used must be verifiably PFCs free.

Another approach, which was used in this application note, is the installation of a delay column between the LC pumps and the injector. The diagram below (Figure 2) demonstrates the proper placement of the delay column in the HPLC fluid path. The interfering species from the solvents and LC pumps will concentrate at the front of the delay column resulting in a shift of retention time. This shift in retention time effectively separates interfering species from the analyte PFCs. This is a simple and costeffective solution.

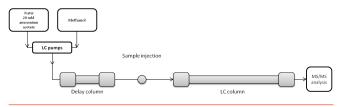


Figure 2
Diagram of delay column installation on HPLC.

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PFOA is an interfering species when analyzing PFCs. Two solutions, a blank consisting of methanol and a 0.5  $\mu$ g/L solution of PFOA in methanol were analyzed (Figure 3) to demonstrate the efficiency of the method using a delay column. (Silact<sup>™</sup> C18 LC-P 50 x 2.1 cm, 3  $\mu$ m)

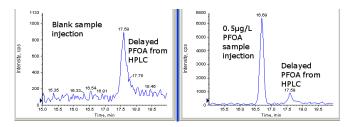


Figure 3 Injection of methanol blank (left) and injection of 0.5 $\mu$ g/L PFOA in methanol (right)

The injection of a  $0.5~\mu g/L$  PFOA solution (Figure 3) shows two peaks. The first one at 16.69 min is PFOA in the injected solution at  $0.5~\mu g/L$ , while the second one at 17.59 min corresponds to the delayed PFOA interference from HPLC lines and solvents. The injection of a blank methanol showed no interference at the expected retention time. The delay column allows the elimination of interfering PFOA at a concentration estimated between  $0.05~\text{and}~0.1~\mu g/L$ .

### **ANALYSIS OF 10 PFAS IN 500 mL OF TAP WATER**

The SPE protocol was carried out on the Gilson 274 ASPEC® Large Volume System in tap water using AttractSPE™ WAX-PFC 6 mL cartridges. The polymeric WAX phase contained in these cartridges undergoes a proprietary treatment at Affinisep to ensure it is entirely PFCs free and a specific high loading capacity for PFCs.

The automation of PFCs analysis was carried out using the Gilson 274 ASPEC® Large Volume System equipped with segregated waste lines and VALVEMATE® II sample selection valve actuator. PEEK tubing was used to reduce risk of PFC adsorption and resulting contamination. The solvents and the 500 mL samples of tap water were contained in Gilson HDPE bottles.



### SAMPLE PRE-TREATMENT

The tap water hardness was very high (Ca²+ ~300 mg/L, NO₃- ~20 mg/L, Cl₂ ~0.3mg/L). For each sample, 475 mL of tap water was mixed with 25 mL of methanol. The pH was adjusted to ~4 with 100  $\mu L$  formic acid. The solution was then spiked with a mix of 10 PFCs at a concentration of 24 ng/L.

One non-spiked sample blank and four spiked samples were processed and analyzed. The blank was analyzed to verify non-contamination.

The analyses were carried using a manual SPE manifold as a control, and on the automated system (Gilson 274 ASPEC® Large Volume System) to confirm the absence of adsorption and contamination potentially introduced by the automated system.

### SPE PROTOCOL

Step	SPE Protocol
Conditioning/ Equilibration	1. 5 mL 0.1% NH <sub>4</sub> OH in methanol 2. 9 mL methanol 3. 9 mL HPLC grade water with formic acid (pH = 4)
Loading	500 mL of loading solution (475 mL tap water + 25 mL methanol pH = 4)
Drying	20 mL air pushed through the cartridge
Elution	1. 2 mL methanol 2. 4 mL 0.1% NH <sub>4</sub> OH in methanol

This protocol was easily transferred to TRILUTION® LH Software for automation:



Figure 4
Gilson TRILUTION® LH SPE method



Figure 5 GX-274 ASPEC® LV Load task

The elution was collected in polypropylene vials, homogenized, and directly analyzed by LC-MS/MS.

To determine matrix effect, a fraction of the elution of the blank water sample was spiked at 2  $\mu g/L$  and analyzed.

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## HPLC/MSMS PROTOCOL

The analytical conditions are presented in the table below:

 Table 2

 LC-MS/MS conditions for the analysis of the 10 PFCs.

LC Conditions	MS Conditions
LC Dionex U3000	Qtrap 4000 ESI- MS/MS
Column: Silact C18 LC-P 150 x 2.1 cm, 3 µm and	Curtain gas: 30
pre-column filter at 30°C Delay column: Silact C18 LC-P 50 x 2.1 cm, 3 μm	CAD: High
Injection volume: 5 μL	IS:-4500 V
T° sampler: 10°C	Temperature: 400°C
Flow rate: 0.25 mL/min	GS1/GS2: 50/50

Time (min)	Solvent A	Solvent B	Analyte	Retention time (min)	Q1	Q3	CE (V)
0	60	40	PFBA	4.5	213.0	168.8	-14
1	60	40	PFPeA	8.6	263.0	218.8	-12
20	10	90	PFBS	9.4	299.0	79.8/98.9	-52/-44
30	10	90	PFHxA	12.2	313.0	268.9/119	-14/-28
31	60	40	PFHpA	14.8	363.0	318.8/168.8	-16/-26
35	60	40	PFHxS	14.9	399.0	79.9/98.9	-74/-56
Solvent A: 20 mM ammonium acetate (in water) Solvent B: methanol		PFOA	16.7	413.1	368.9/168.8	-14/-26	
		PFOS	18.3	499.0	80.1/98.9	-84/-70	
		PFNA	18.3	463.0	418.9/219.0	-16/-24	
		PFDA	19.6	513.0	469.0/218.8	-13/-11	







Table 3
Recovery of 10 PFCs in 500 mL of unspiked tap water and spiked tap water and observed matrix effect after purification with AttractSPE™ WAX-PFC on Gilson 274 ASPEC® Large Volume System.

Compound	Unspiked Tap Water	Spiked Tap Water at 24 ng/L	RSDr (n=4)	Blank water Eluate spiked at 2 µg/L
		Recovery		Observed matrix effect
PFBA	ND*	99%	3%	+16%
PFPeA	ND*	99%	1%	+12%
PFBS	ND*	101%	3%	+10%
PFHxA	ND*	102%	3%	+17%
PFHpA	ND*	100%	4%	+8%
PFHxS	ND*	101%	2%	+1%
PFOA	ND*	102%	2%	-1%
PFOS	ND*	87%	2%	+8%
PFNA	ND*	97%	3%	+8%
PFDA	ND*	83%	2%	-1%

\*ND: Not Detected

### **RESULTS**

The method linearity was checked against a calibration curve with PFCs concentrations of 0.5, 1, 2, 4, and 6  $\mu g/L$ . The  $R^2$  value was found to be greater than to 0.998 for the ten molecules, demonstrating linearity of the method over this concentration range.

As shown in the Table 3, no detectable contamination was found using the AttractSPE™ WAX-PFC cartridges on the Gilson 274 ASPEC® Large Volume System.

The elution fraction of a blank sample was spiked at 2  $\mu$ g/L showing **no significant matrix effects**, with a maximum of 16% signal enhancement for PFBA by comparison with the calibration curve. Without an additional concentration step after the SPE protocol, good values of LOQ (6 ng/L) and LOD (2 ng/L) were achieved.

Futhermore, AttractSPE™ WAX-PFC showed excellent recoveries from 83% to 102% and the Gilson 274 ASPEC® Large Volume System allowed excellent relative standard deviation from 1% to 4%.

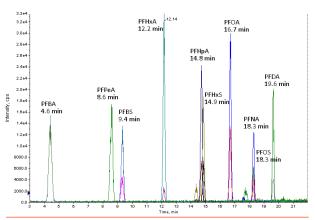


Figure 6
LC-MS/MS chromatogram for the 10 PFCs (see HPLC/MS

### CONCLUSION

The automation of the isolation of 10 PFCs from tap water for analysis by LC-MS/MS, was found to be very effective with good recoveries from 83% to 102% without any contamination from the extraction system.

The AttractSPE™ WAX-PFC cartridges have enhanced selectivity and concentrated the sample more than 80 fold, allowing the direct analysis of the 10 PFCs with a LOQ at 6 ng/L without any evaporation step.

The combination of AttractSPE™ WAX-PFC and the Gilson 274 ASPEC® Large Volume System allowed development of a robust method with excellent repeatability (RSD 1%-4%) that helps eliminate environmental variables and effectively reduces chances of sample contamination.

Furthermore, the automation greatly reduces labor cost, reduces potential exposure to hazardous solvents, and increases overall efficiency for high throughput laboratories.

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### PART NUMBERS OF PRODUCTS USED IN THIS APPLICATION NOTE

CONSUMABLES			
Part Number	Product Name	Quantity	
WAX-PFC-50.S.6	AttractSPE™WAX-PFC 6 mL	50/pk	
C18LCP-150.2.1	HPLC column: Silact C18 LC-P 150x2.1cm 3μm	1 unit	
C18LCP-50.2.1	HPLC delay column: Silact C18 LC-P 50x2.1cm 3μm	1 unit	

GILSON 274 ASPEC LARGE VOLUME SYSTEM		
Part Number	Product Name	Quantity
2614010LV	GX-274 ASPEC, LARGE VOLUME, WITH Z DRIVE	1

### Reference

<sup>1</sup>Method 537.1: Determination of Selected Per- and Polyfluorinated Alkyl Substances in Drinking Water by Solid Phase Extraction and Liquid Chromatography/Tandem Mass Spectrometry(LC/MS/MS). https://cfpub.epa.gov/si/si\_public\_record\_Report.cfm?dirEntryId=343042&Lab=NERL

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